Deuterium excess anomaly of precipitation in Svalbard



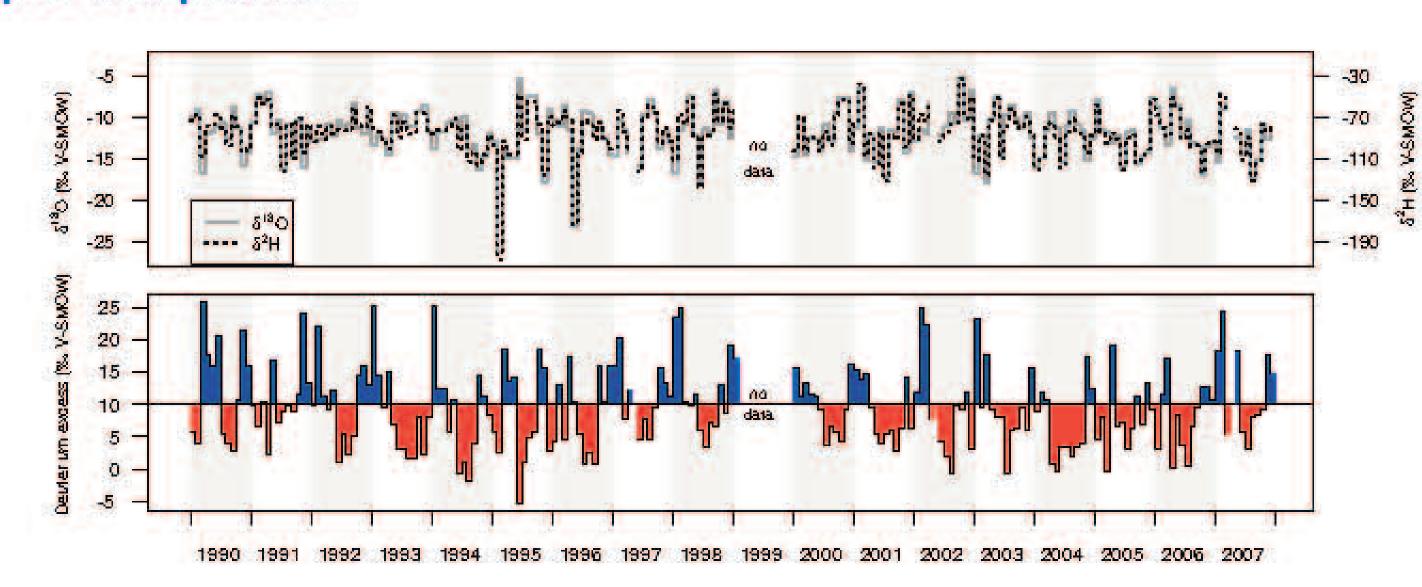
Matthias Ritter (1), Christoph Kuells (2)

Precipitation

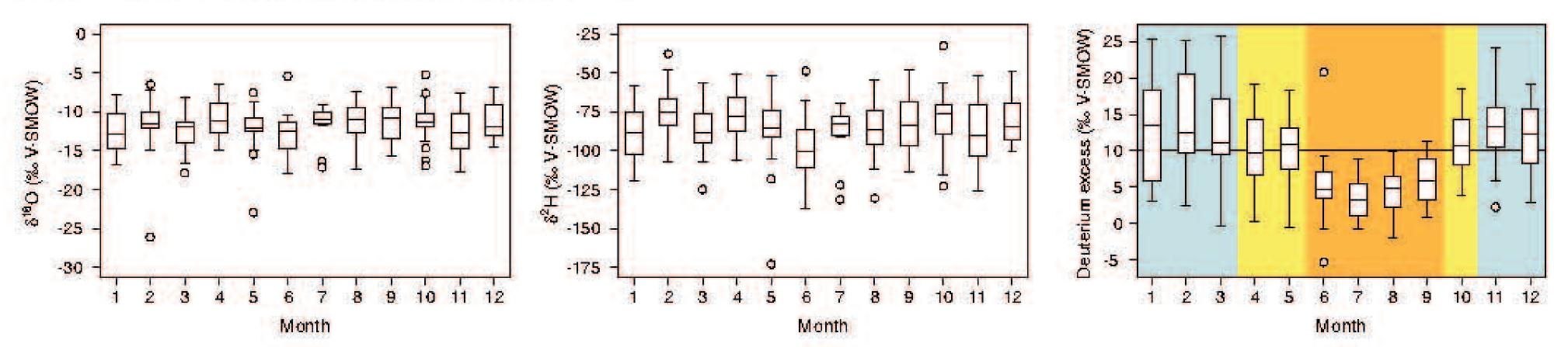
Temporal variation of isotopic composition

Time series of monthly isotope ratios and deuterium excess in precipitation at IAEA GNIP station Ny Alesund $(78.15^{\circ}\text{N}, 11.56^{\circ}\text{E}, 7\text{m})$. Axes of $\delta^2\text{H}$ and δ^{18} O are scaled to match for a deuterium excess of 10 (=GMWL).

Deuterium excess values >10 are indicated by blue, <10 by red bars.



Seasonality of isotopic composition

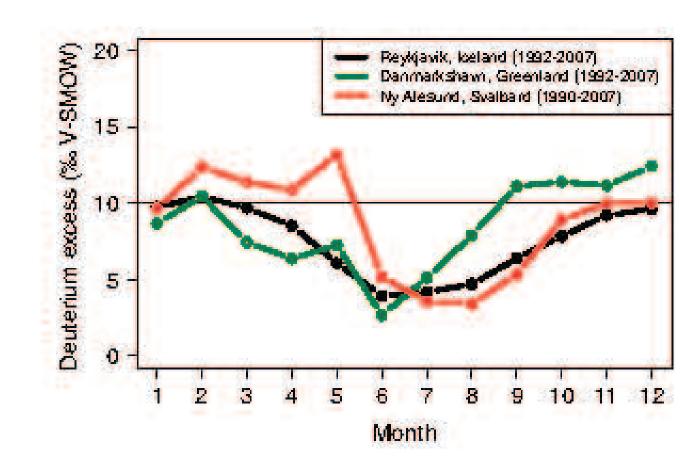


Isotope ratios δ^2H and $\delta^{18}O$ are influenced by local atmospheric conditions during precipitation formation. Thus they represent single event characteristics and follow no clear seasonal pattern. However, a strong seasonality is obvious for deuterium excess, suggesting a winter (blue), transition (yellow) and

summer (orange) period. It reflects differences in the moisture source area, as the isotopic content of lower-tropospheric vapour over Svalbard exhibits the same pattern (Frankenberg et al., 2009) and secondary processes (sub-cloud evaporation, snow formation) seem to be of only minor importance.

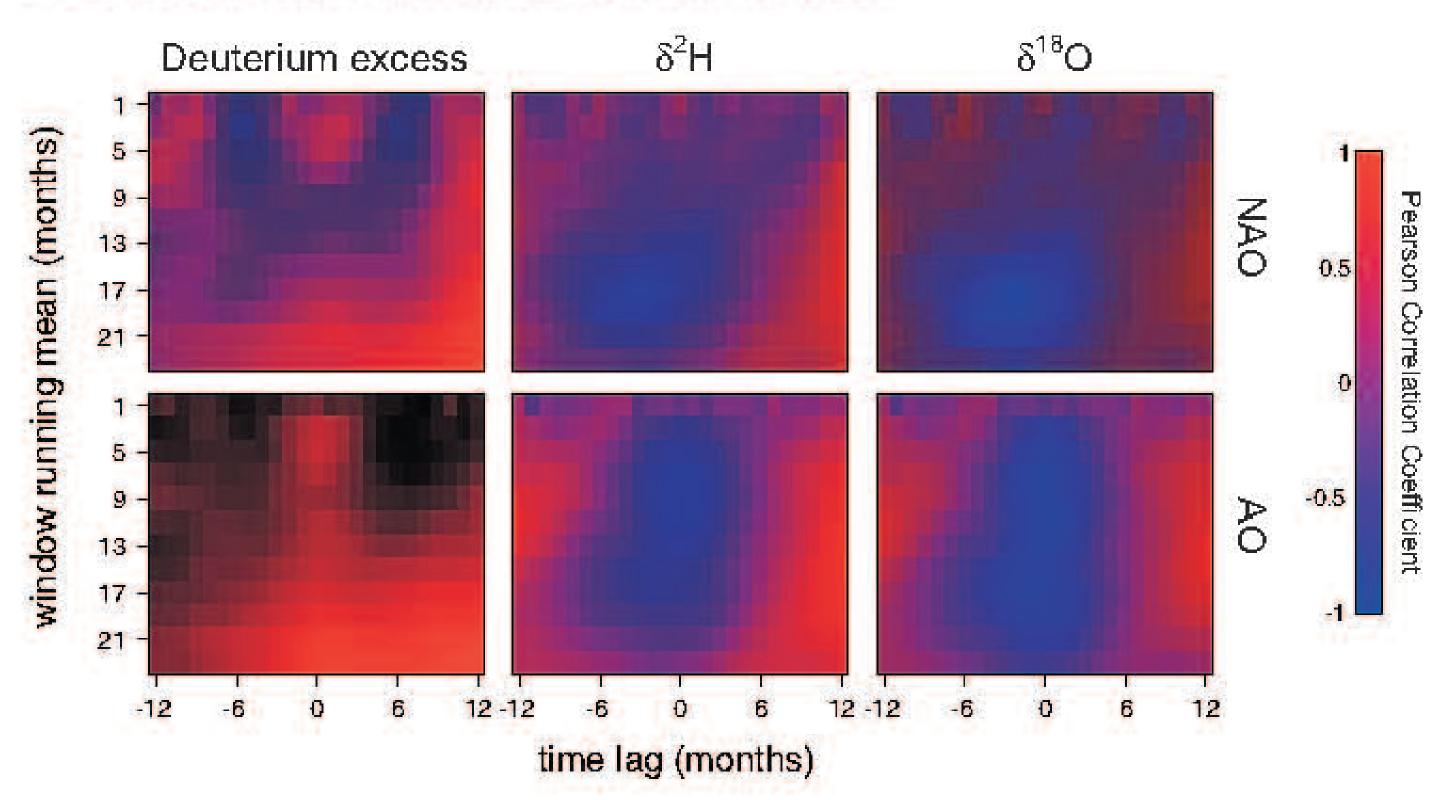
Comparison with other high latitude stations

Other high latitue GNIP stations in Greenland and Iceland also have a pronounced annual pattern in deuterium excess with a smaller amplitude in Iceland. Winter values are related to a strong kinetic effect during evaporation from local source regions with large sea surface - air temperature contrasts.



During the summer period more atmospheric moisture originating from the warmer North Atlantic intrudes into the polar regions. Hence, vapour is formed under conditions that are characterized by less kinetic fractionation. This contributes to the yearly minimum deuterium excesses.

Coupling with atmospheric circulation



Cross correlation analyses with the North Atlantic Oscillation (NAO) and Arctic Oscillation (AO) indices show noticeable differences between deuterium excess and the iotope ratios on different time scales. The results highlight the importance of large scale atmospheric circulation controlling moisture fluxes from different regions. As long-term trends in circulation indices are well known, effects on the isotopic composition of precipitation can be expected. However, time series of isotopic precipitation composition in Svalbard do not show visible trends (yet).

Background

Stable water isotopes (²H, ¹⁸O) are a widely used tool for fractionation processes, the deuterium excess contains also environmental questions.

The isotope ratios $\delta^2 H$ and $\delta^{18} O$ in precipitation mainly depend

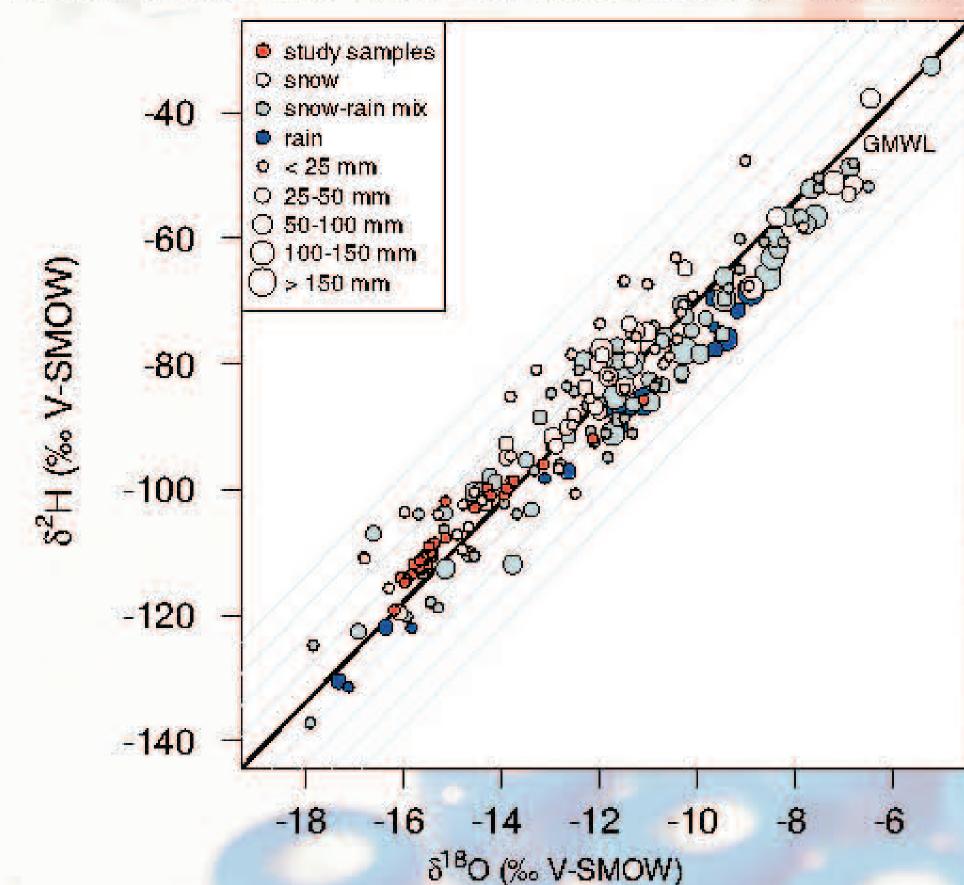
environmental and dimate system analyses. During the last information about kinetic (non-equilibrium) isotope years new measurement techniques (Cavity Ring-Down fractionation. Especially influenced by ocean-atmosphere Spectroscopy) facilitated the joint quantification of both δ^2H interactions during evaporation, deuterium excess can be and δ^{18} O. Consequently, the second-order isotope parameter—used to delimit spatially and temporally variable moisture deuterium excess (δ^2 H - 8 δ^{18} O) can be utilized to study—source regions. The contrast between sea surface and air temperature and the humidity and wind profile at the evaporation site define the anomaly of deuterium excess. on temperature conditions during precipitation formation and However, there is still no unified theory which explains the

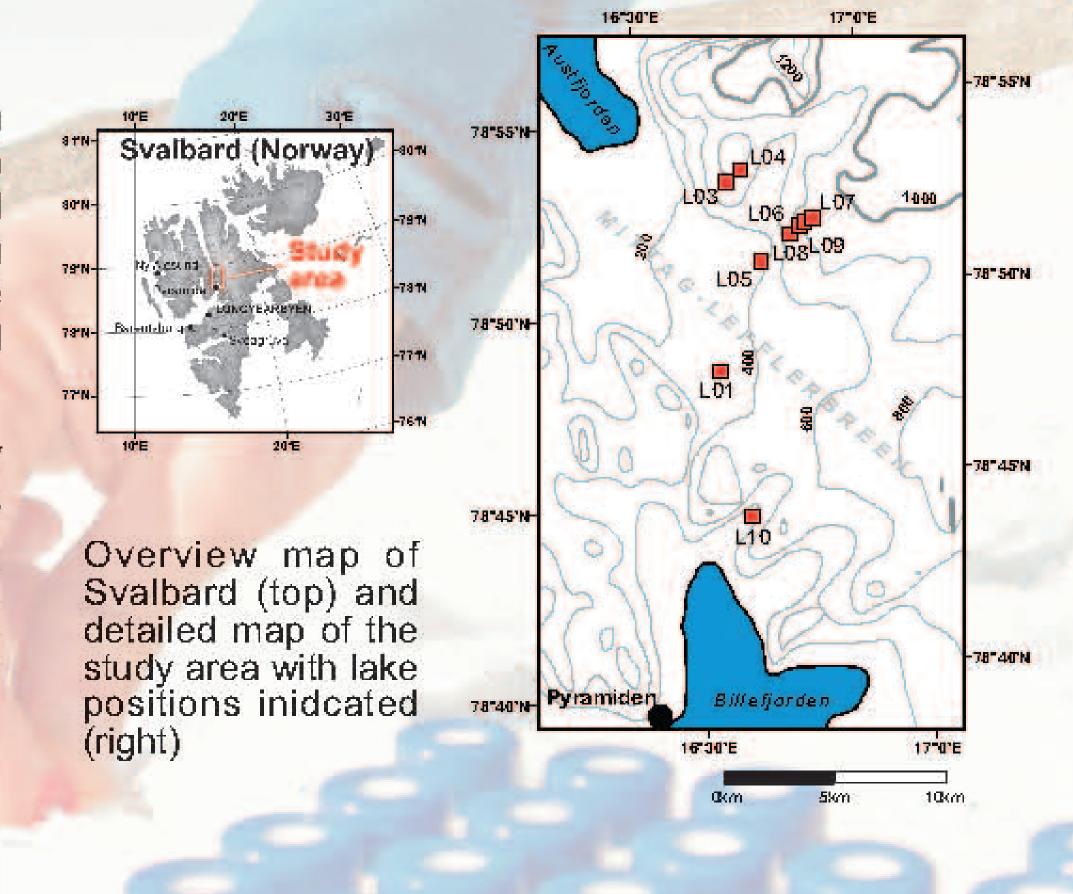
local characteristics such as latitude, continentality and generation processes and the spatio-temporal distribution elevation. Whereas these are controlled by equilibrium—of deuterium excess in atmospheric vapour and precipitation.

Study area and data

During a field campaign in August 2009 water samples from ten high-arctic lakes adjacent to the Mittag-Lefflerbreen glacier on Svalbard were taken. Additionally, inflowing and outflowing surface waters and the top layer of the surrounding snow pack were sampled. For each lake the outflow rate and surface area (using a handheld GPS unit) were estimated in the field.

Laser spectrometry (Picarro G1102-i) was used to quantify the δ^2 H (±1‰) and δ^{18} O (±0.1‰) isotope ratios of the samples.





Most samples plot parallel to the global meteoric water line (GMWL) with deuterium excesses of ~13.5. Smaller deuterium excess values are observed for samples with δ^2H larger than

The isotope ratios are within the range of the isotopic composition of precipitation at the coastal IAEA GNIP station located in Ny Alesund (1990-2007).

Conclusions

Deuterium excess of precipitation in Svalbard shows a strong Deuterium excess of precipitation is reflected by surface

seasonal signal due to different moisture source areas. waters and can be traced through the hydrologic cycle.

The seasonal signal is linked to circulation patterns and The studied lakes are dominated by direct snow melt water, provides a possibility to detect changes in vapour sources but diffuse contribution of subsurface water can be substantial. and atmospheric processes.

With the nowadays common combined analysis of both δ^2 H and δ^{18} O the deuterium excess is a useful indicator to characterise hydrological processes and systems, especially in high latitude regions. However, the mechanism of deuterium excess generation must be explored for a complete understanding of its information content.

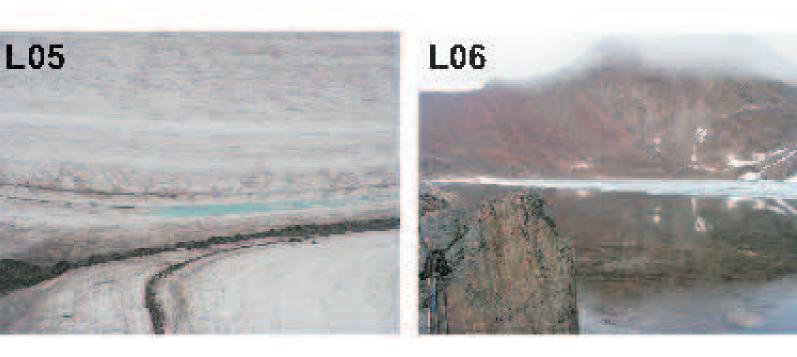
Lakes

Lake characteristics

The studied lakes cover an elevation range of 600m and feature differences in mapped water sources. Despite high uncertainties due to unknown mean lake depths, the calculated mean residence times (MRT) indicate a complete tumover of lake water within days to few weeks for most lakes.

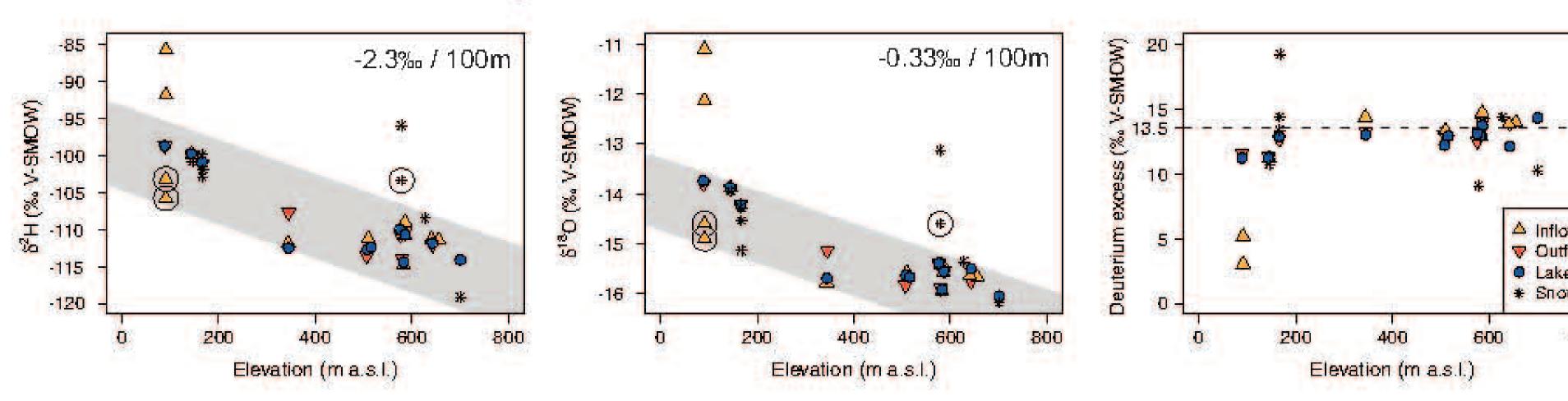






Lake	Description	Elevation (m a.s.l.)	Watersource	Lake area *) (m²)	Lake volume ⁶⁾ (m²)	Outflow ⁴ (m¾d)	MRT (d)
LO1	Framstakken (lower lake)	145	upper lake (L02), snow	560 ~112)	3584 (±3136)	168(±51)	41 (±39)
102	Framstakken (upper lake)	166	permatrost, subsurface, snow	3900 (±780)	24960 (±21840)	216 (±65)	222 (±211)
ша	Sedemalmfeller (lawer)	580	upper lake (L04), snow	26000 (±5200)	166400 (± 145600)	4500 (±1350)	71 (±68)
104	Sederhalmfeller (upper)	701	permatrost, snow	1600 (±320)	10240 (±8960)	no data	no data
W5	Gladier lake Camp	345	glacier	5300 (±1060)	33920 (±29680)	4985 (±1495)	13 (±12)
106	Bonfeller 1 (lower)	429	upper lake (L07), snow	130000 (±26000)	832000 (±728000)	9874 (±2962)	162 (±154)
L07	Bonfeller2	576	upper lake (L08), snow	1600 (±320)	10240 (±8960)	14698 (±4406)	1.3 (±1.2)
lO3	Bonfeller3	587	upper lake (L09), snow	7600 (±1520)	48640 (±42560)	21600 (-6480)	4.3 (54.1)
109	Bonfeller 4 (upper)	643	snow, glader	3700 (±740)	23680 (±20720)	21600 (-6480)	2.1 (±2.0)
Lta	Ragnarmoraine left	89	moraine	4400 (±880)	28160 (±24640)	71 (±21)	766 (±728)
				uncertainty +20%	uncertainty mean lake elepth 1 10m	uncertainty +30%	

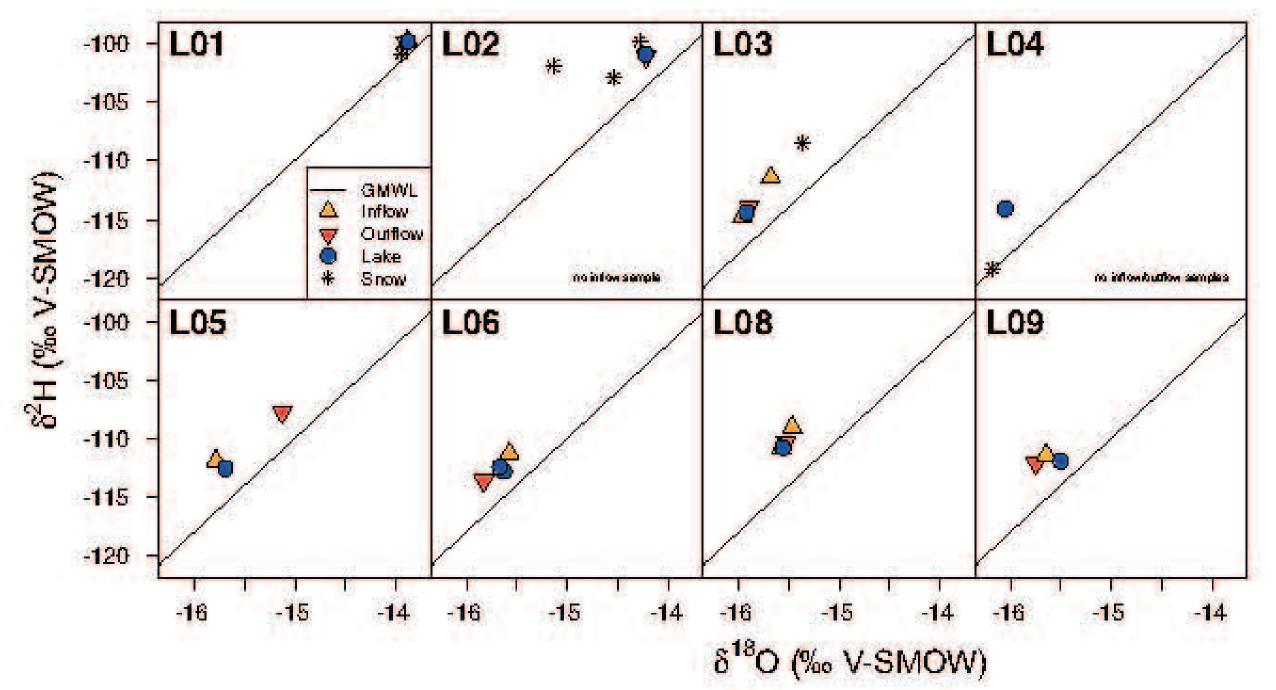
Elevation effect of water samples

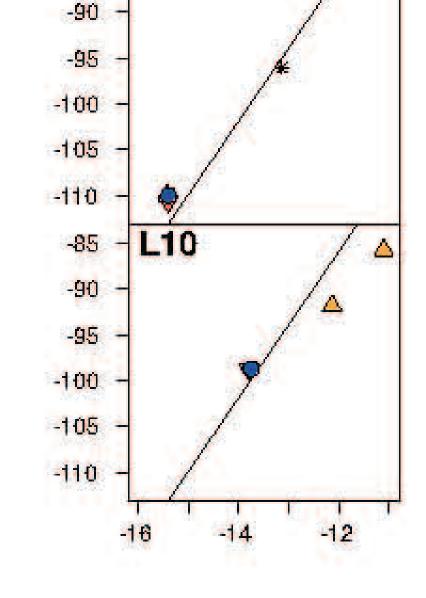


Both δ^2 H and δ^{18} O values show a depletion with increasing elevation due to stronger fractionation at lower temperatures. The observed gradient for δ^{18} O is close to the theoretical value of -0.36% / 100m for adiabatic cooling. Enriched samples were corrected to a deuterium excess of 13.5 (indicated by circles).

Deuterium excess is ~13.5 for the high elevation lakes and reflects winter precipitation. It is lower for the water sampled below 200m. This suggests evaporative enrichment at the lowest sites as a consequence of the earlier onset of snow melt and possibly longer flowpaths near the surface or a local contribution of summer precipitation.

Inferring lake water soures





Lake water was sampled bankside and was not always well mixed (L05, L06, L09).

Snow samples are highly variable in their isotope composition and probably influenced by evaporation and sublimation after deposition.

Most liquid water samples plot above the GMWL and contain the signature of winter precipitation (deuterium excess > 10). This demonstrates the dominance of water input from snow melt. The inflows to the lowest lake (L10) are isotopically enriched. To explain lake water isotopic ratios, significant subsurface inflow is necessary. However, this groundwater flow must be sustained by winter precipitation to achieve the lake water composition. Subsurface inflow is also likely for other lakes (L03, L06, L08) and the glacier lake L05 (englacial

Acknoledgement:

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