

An isotopic comparison of waters obtained by destructive and non-destructive methods to evaluate mixing and runoff processes at the mini-hillslope scale

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Introduction

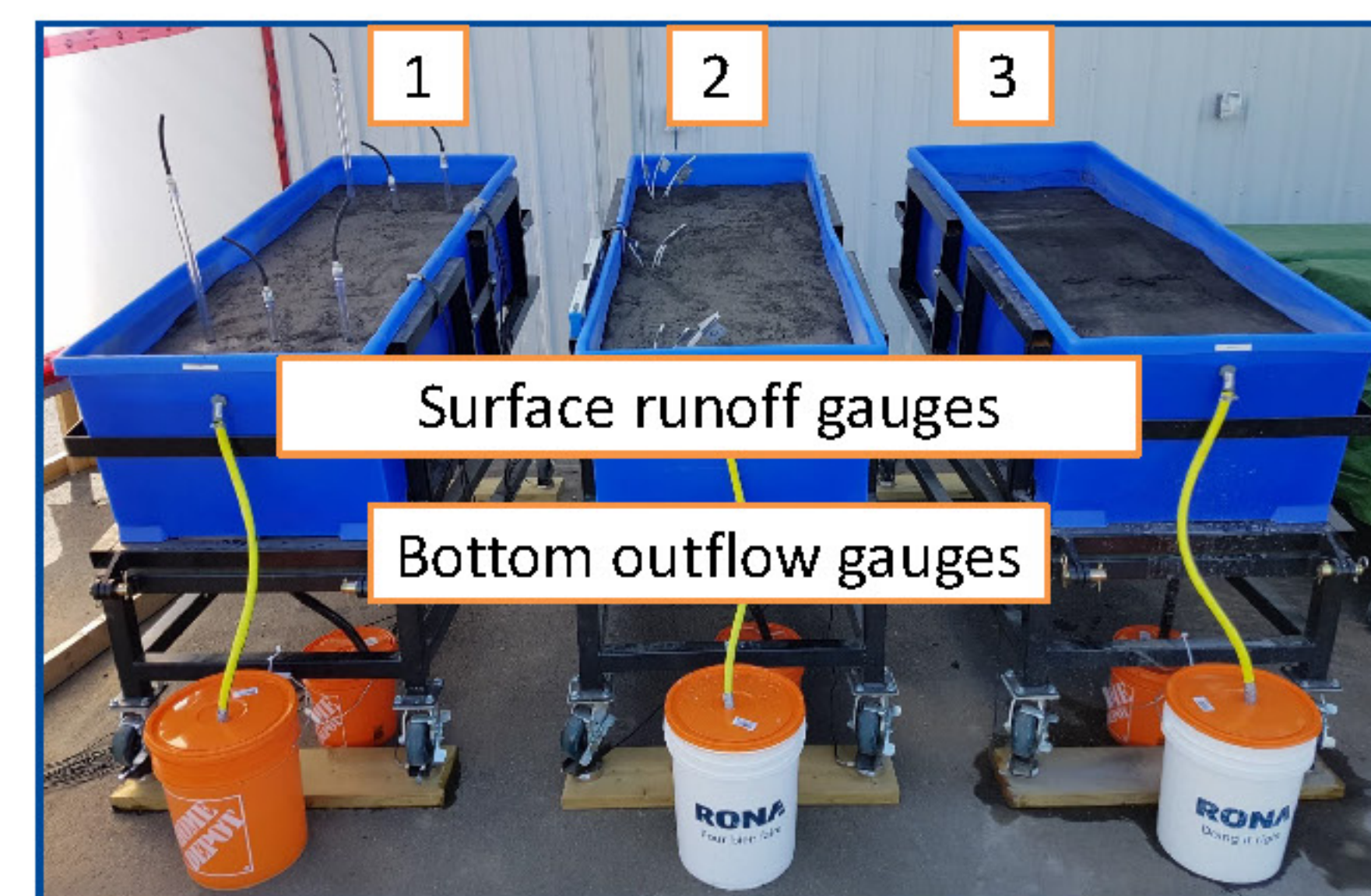
- Technical developments have fostered the application of stable water isotopes (^2H and ^{18}O) to measure, understand and predict water flow paths and different pools of subsurface water
- To gain the isotopic information preserved in the soil, many pore water extraction methods exist, but their effects on natural soil water isotopic composition are poorly understood
- Findings have major consequences for the use of a soil water component for hydrograph separation, mean transit time estimation or plant water uptake depth estimates

Null hypothesis:

Destructive and non-destructive extraction methods sample isotopically the same soil water pool

We intercompared destructive and non-destructive water extraction methods on an outdoor mini-hillslope:

- Non-destructive: Suction cups (Slope 1) and in-situ vapour ports (Slope 2)
- Destructive: Centrifugation and vapour equilibration method (Slope 3)



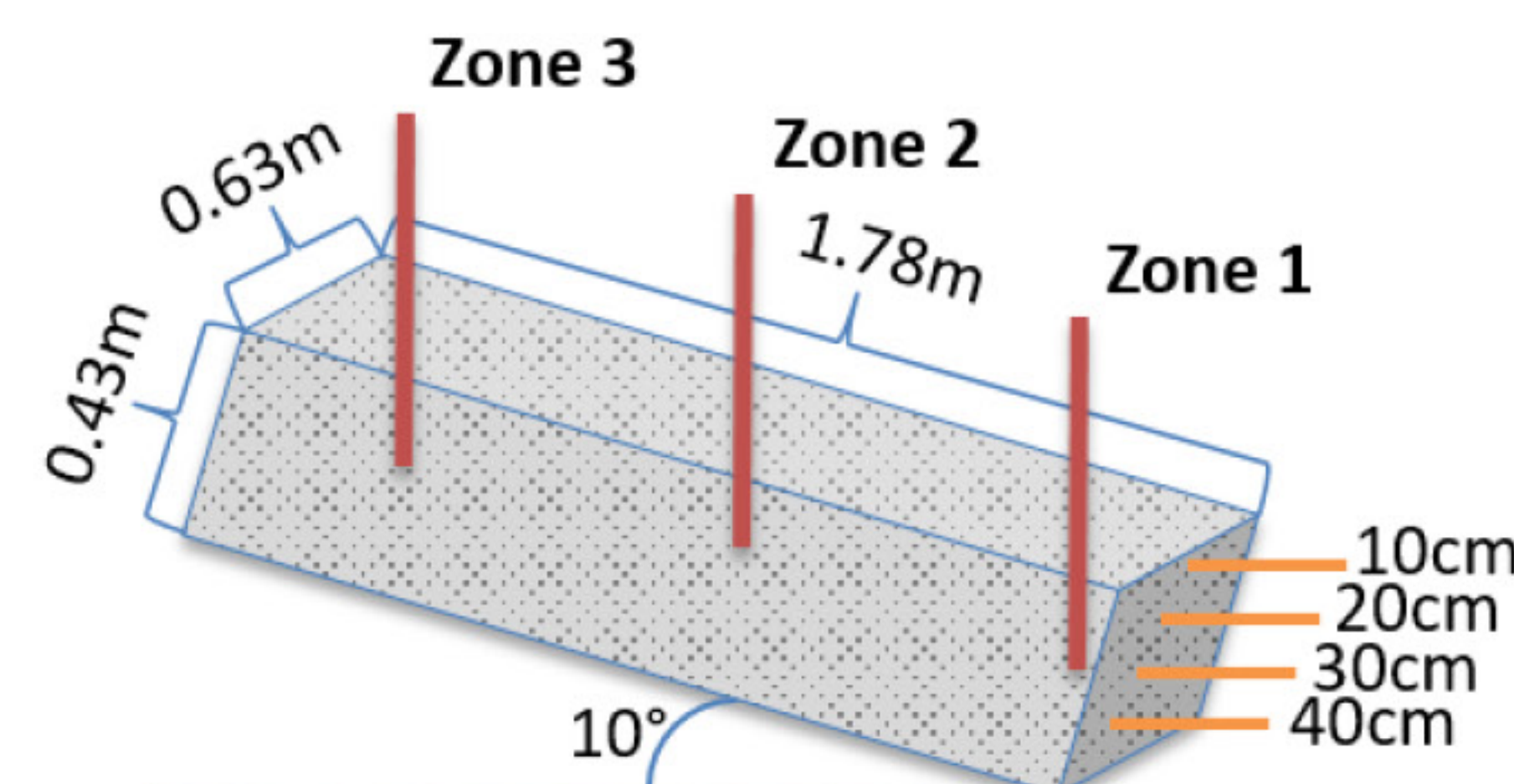
Materials and Methods

- Hillslopes contain volcanic basalt rock of loamy sand texture (Biosph.2; U of A, USA)
- Slopes initially filled with tap water
- Afterwards, natural precipitation used as input function to study water flow paths isotopically over and through the soil profile of the hillslopes
- Local Meteoric Water Line (LMWL) from regression analysis of local precipitation data
- Is-excess as indicator for non-equilibrium fractionation:

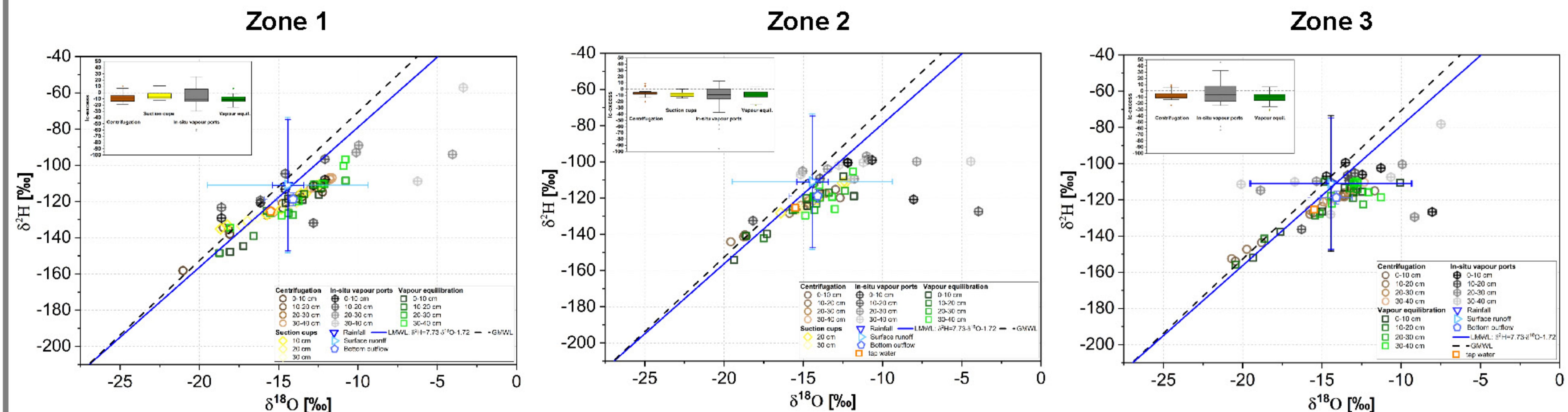
$$lc - excess = \delta^2H - \delta^{18}O * a - b$$

with a=slope and b=intercept

- Measurement equipment: Climate station, soil moisture & temperature sensors, load cells under each hillslope, bottom outflow and surface runoff gauges
- 6 non-/destructive samplings; 3 replicates per depth and zone
- ^2H and ^{18}O analyses on a laser spectroscope (IWA-45EP Analyzer, LGR Inc., US; precision: $\pm 0.5\text{‰}$ for $\delta^2\text{H}$ and $\pm 0.1\text{‰}$ for $\delta^{18}\text{O}$ for liquid water; $\pm 0.2\text{‰}$ for $\delta^2\text{H}$ and $\pm 0.05\text{‰}$ for $\delta^{18}\text{O}$ for water vapor)

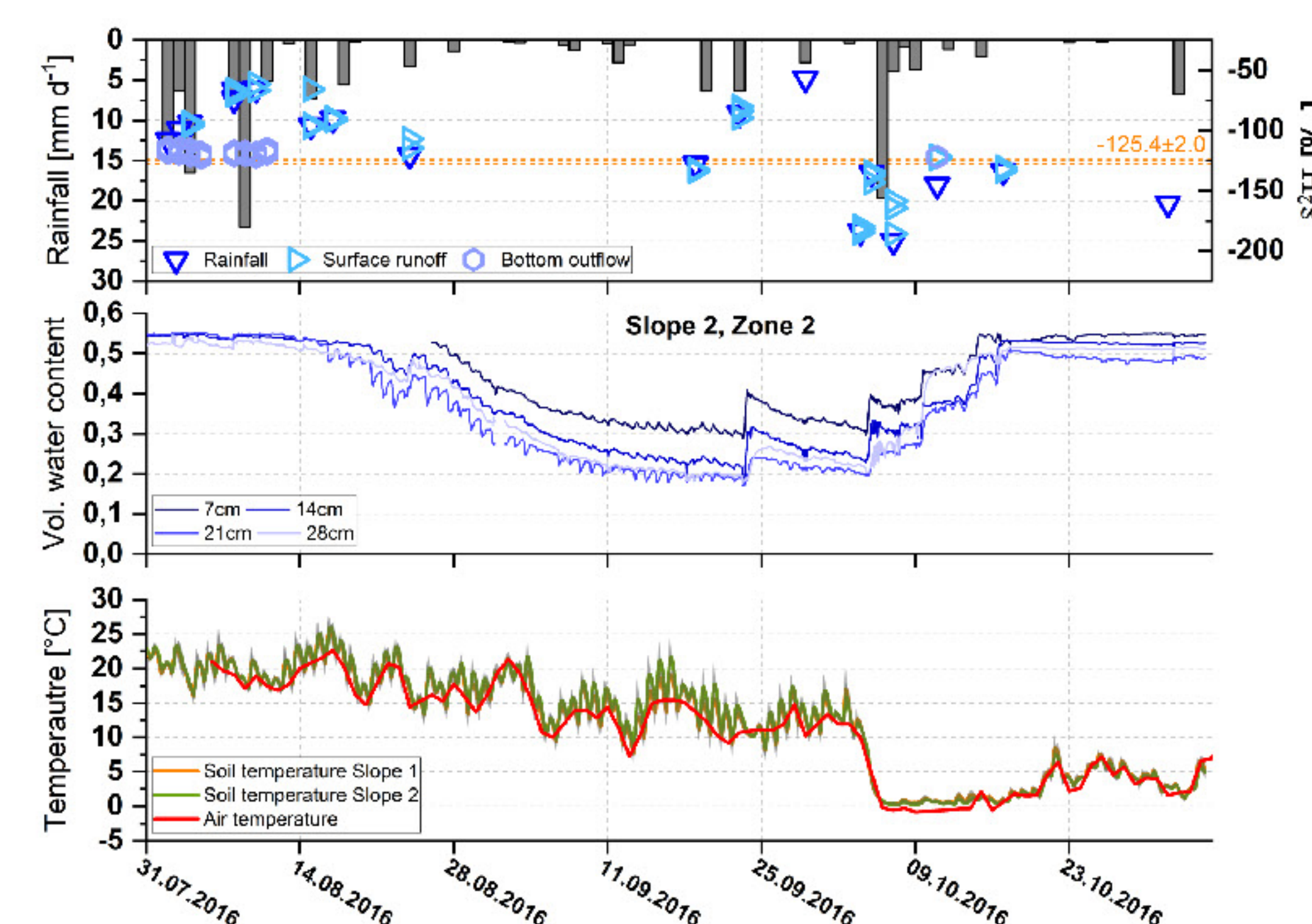


Results and Discussion



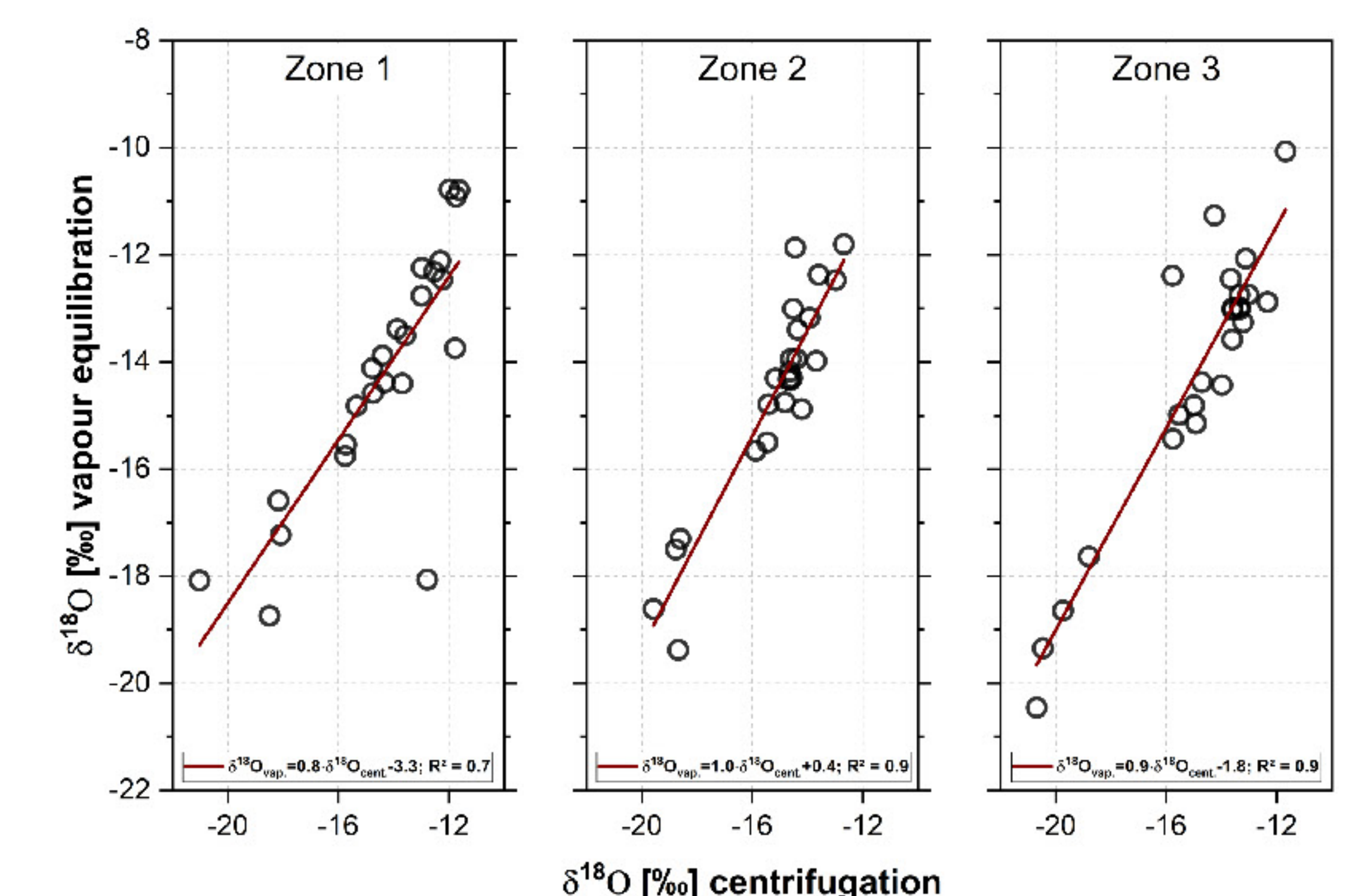
Dual isotope data

- All methods' isotope results fell within the range of precipitation and runoff, except for the in-situ vapour ports
- Isotope results from the in-situ vapour ports showed the greatest offset from the LMWL and the input signal, especially for the lowest depth (30-40cm)
- Issues with in-situ vapour ports were most likely due to high water contents causing condensation in the tubing
- Suction cups failed when soils were dry and pores most likely became clogged with soil material
- Centrifugation and vapour equilibration method plotted in the same isotopic range for all hillslopes and even showed a 1:1 relationship
- Destructive and non-destructive extraction methods did not sample the isotopically same soil water pool



Temporal variation

- Precipitation was isotopically more enriched in summer than in autumn as air temperatures dropped below zero
- Surface runoff followed the isotopic trend of the precipitation input and fell on the LMWL
- Bottom outflow was not always present and was isotopically more depleted
- Soil temperatures of both slopes were almost identical and followed air temperature trends



Conclusions

- Centrifugation and vapour equilibration were most reliable
- Applied extraction methods most likely sampled the more mobile soil water pool as results fell within the range of precipitation
- Aim of individual study dictates which method is suitable for particular results, sample types and properties, timeline, cost, and precision

